Stereoselective Synthesis of N¹-Lyxitol Inosine Derivative

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Abstract: 1,4-anhydro-2-triflyl-3,5-O-benzenylidene-L-xylitol (5) was constructed in six steps from protected D-xylose. Substitution of 5 with protected 8-bromoinosine 6 gave the key intermediate 5'-O-TBS-2',3'-di-O-acetyl-N¹-(2"-deoxy-1",4"-anhydro-3",5"-O-benzenylidene-L-lyxitol-2"-yl)-8-bromoinosine (14). Selective removal of 5'-O-TBS-group gave the corresponding 5'-O-phosphorodianilidate 4 though phosphorylation, which was characterized by X-ray crystallographic analysis.

Keywords: Cyclic ADP-ribose analogues, synthesis, X-ray crystallographic analysis.

Cyclic ADP-ribose (cADPR) **1** (**Figure 1**), a intracellular Ca^{2+} -mobilizing messenger, is of great interest because of its significant importance in the signal transduction¹. However, the biological as well as chemical instability of cADPR limits the studies of its physiological role, at least in some aspect². Recently, Shuto *et.al.*³ constructed a mimic of cADPR, cyclic IDP-carbocyclic-ribose **2**, in which the 4'-oxo in furanose is substituted by a methylene group and the stability of **2** is improved.





Isonucleosides represent another class of nucleoside analogues in which the nucleobase is linked at various positions of ribose other than C-1', and have attracted great interest because of their chemical and enzymatic stability⁴. In order to study the relationship between the structure and biological activities, we designed a stable cADPR mimic with glycosyl instead of ribosyl in the N-1 position.

The key steps for the synthesis of cADPR analogue **3** involve glycosidation at N-1 position of inosine and intramolecular cyclophosphorylation. We reported here the synthesis of N^1 -lyxitol inosine derivative **4**.

Li Jun HUANG et al.

It was reported that N-1 substitution of hypoxanthine can be carried out *via* a rearrangment of purine⁵, but we failed to construct the key intermediate by this method. The synthesis of N¹-lyxitol inosine derivative was completed by the nucleophilic substitution with compound **5** and **6**.



a. TsCl,;Py, rt; b. 1% HCl, MeOH, 80°C; c. 1% HCl, H₂O, 90°C; d. NaBH₄, rt; e. ZnCl₂, PhCHO, rt; f. $(CF_3SO_2)_2O$, Py, DCM, -20°C; g. TBSCl, imidazole, DMF; h. $(AcO)_2O$, Py, rt

The preparation of protected xylitol **5** and inosine was shown in **Scheme 1**. Because of the low acidity of the N¹-H in compound **6**, triflyl was selected as leaving group at 2'-position of compound **5** for the nucleophillic substitution. Compound **11**, 3,5-O-benzenylidene-1,4-anhydro-L-xylitol, was obtained from **7** using the method previously reported in five steps⁶. Addition of trifluoromethanesulphonic anhydride to the mixture of **11** and anhydrous pyridine in DCM at -20°C under argon and stirring for one hour at the same temperature afforded triflate **5** which was unstable and used for the next reaction immediately without purification. 8-bromoinosine (**12**) was synthesized from adenosine according to a publication procedure⁷. A selective protection of 5-primary hydroxyl was accomplished by the treatment of **12** with t-butyldimethylsilyl chloride (TBS) under optimized condition. The desired inosine derivative **6** was obtained in over 90% yield from **12**.

As depicted in Scheme 2, a suspension of 6, K_2CO_3 and 18crown-6 was heated under reflux for one hour, and then cooled to room temperature. Equimolar 5 in THF was added to the mixture. Upon further stirring at 40°C for 36 hours, the expected product 14⁸ was isolated by column chromatography with the yield of 24%, accompanied by O⁶-substituted derivative 16 in 6.8% and eliminated byproduct 15. The 5'-O-TBS group of 14 was removed with TBAF in neutral THF solution to give the 5'-hydroxyl derivative 17 and 17 was treated with (PhNH)₂POCl and tetrazole in pyridine to give 4⁹ in high yield (Scheme 2).

Scheme 2



d. (PhNH)₂POCl, tetrazole, Py, rt.

The N1-glycosidic linkage of inosine was confirmed by IR, UV and NMR. The IR spectrum of **14** showed a strong v c=0 absorption peak at 1706 cm⁻¹. Similar absorption around 1700 cm⁻¹ was not found in the IR spectrum of 16. Correspondingly, UV spectrum of 14 has a lower value of A260/A290 than its O⁶-substitued counterpart owing to the absence of conjugation.

Figure 2



In order to determine the stereochemistry of 14, the single crystalline of 4 was obtained from ethanol solution. The crystal belongs to tetragonal system and space group is $P4_3$ (Figure 2). The detail of unit-cell parameters was a=16.666(1), c=16.987(1)Å. X-ray crystallographic analysis demonstrated that inosine moiety was on the same

side as the benzenylidene group. It was deduced that the absolute configuration of C-2" is S configuration as same as C-3" and C-5" in compound 14. The result supported an S_N2 substitution mechanism in the substitution that the nitrogen anion attacked the C-2" from the back and gave the configurational reversal product.

Acknowledgments

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References and Notes

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- Compound 14 mp:106-108°C. FAB-MS: m/z=749.2 [M⁺]; 300 MHz ¹HNMR (CDCl₃): δ
 8.37 (s, 1H, H-2), 7.37 (m, 5H, C₆H₅), 6.27 (t, 1H, J=5.7Hz, H-2), 5.97-6.05 (m, 2H, H-1, H-2[°]), 5,70 (t, 1H, J=5.4Hz, H-3), 5.50 (s, 1H, PhCH), 4.60 (m, 1H, H-3[°]), 4.42-4.47 (m, 1H, H-1[°]), 4.27-4.34 (m, 2H, 2H-5[°]), 4.06-4.22 (m, 2H, H-4, H-1[°]), 3.97 (m, 1H, H-4[°]), 3.90 (dd, 1H, J_{5'a,5b}=11.1Hz, J_{5'a,4}=4.5Hz, H-5a), 3.80 (dd, 1H, J_{5'b,5'a}=11.1Hz, J_{5'b,4}=4.8Hz, H-5b), 2.02, 2.13, (each s, each 3H, 2CH₃CO), 0.82, -0.00, (each s, each 3H, (CH₃)₃C-); ¹³C NMR (CDCl₃): δ 169.4, 169.1, 155.5, 148.1, 146.8, 136.9, 129.1, 128.5, 126.2, 125.7, 124.6, 99.1,88.0, 83.1, 74.2, 73.8, 71.6, 70.3, 69.8, 67.4, 62.3, 54.3, 25.7, 20.6, 20.3, 18.3, -5.5, -5.4. Anal Calcd for C₂-H₂-BrN₀, O₂Si[°] C 51.27 H 5 51. N 7.47. Found: C 51.60 H 5 60 N 7.67
- Anal.Calcd. for $C_{32}H_{41}BrN_4O_{10}Si: C 51.27, H 5.51, N 7.47;$ Found: C 51.60, H 5.60, N 7.67. 9. Compound **4** mp:146-148°C. 500 MHz ¹HNMR (DMSO-d₆): 8.22 (s, 1H, H-2), 8.04 (d, 1H, $J_{PH}=3.5Hz$, NH), 8.06 (d, 1H, $J_{PH}=3.0Hz$, NH), 7.33-6.71, (m, 15H, Arom-H), 6.03, (dd, 1H, $J_{1'2}=4.0Hz$, H-1'), 5.80 (t, 1H, H-3'), 5.58 (s, 1H, PhCH), 4.63 (dd, 1H, $J_{3''4}=5Hz$, $J_{3''2}=2.0Hz$, 3"-H), 4.44 (dt, 1H, $J_{4'5a}=6.5Hz$, J=6.5Hz, z, $J_{4'5b}=3.5Hz$, $J_{4',3}=6.5Hz$, H-4'), 4.33 (m, 1H, 5'-H), 4.17-4.26 (m, 4H, H-1'', H-5', 2H-5''), 4.09 (t, J=10.5Hz, 1H, H-1''), 3.88 (m, 1H, H-4''), 2.08, 2.09 (each s, each 3H, 2CH₃CO); Anal.Calcd. for $C_{38}H_{38}BrN_6O_{11}P: C 52.73$, H 4.42, N 9.71; Found: C 52.40, H 4.36, N 9.39.

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